

## Structure prediction in the Zn-Al-O phase diagram

Kim Rijpstra\*, Stefaan Cottenier, Michel Waroquier, Veronique Van Speybroeck

Center for Molecular Modelling, Ghent University, Technologiepark 903, 9052 Zwijnaarde, Belgium

\*K. Rijpstra, Email: kim.rijpstra@ugent.be

In a world that is using ever more transparent conducting oxides (TCO), the ideal  $\text{Sn}_2\text{O}_3:\text{In}$  (ITO) is becoming more and more expensive. A commercially available alternative is  $\text{ZnO}:\text{Al}$  (2 wt%), which is, however, far from optimal. As part of a larger project, our goal is to search and evaluate alternative TCO's.

Since the Zn-Al-O system already delivered a TCO, one could wonder whether more is to be found in this ternary system. This phase diagram has only 3 known binary structures ( $\text{ZnO}$ ,  $\text{ZnO}_2$ ,  $\text{Al}_2\text{O}_3$ ) and 1 known ternary ( $\text{ZnAl}_2\text{O}_4$ , gahnite) [1]. A machine learning method [2] returned a few metastable aluminium-oxides and 1 metastable ternary structure on top of the experimentally known ones. Several low-density ZnO-polymorphs have been predicted ([3], [4]), which should be stable at slight negative pressures. This suggests that metastable ternary Al-Zn-O alloys are not unfeasible.

By making total energy DFT calculations, using VASP [5], for a set of crystal structures that appear experimentally in chemically similar ternary systems, we identified 16 new Al-Zn-O alloys. They form a line in the Al-Zn-O composition triangle, connecting ZnO and  $\text{Al}_2\text{O}_3$ . Every composition on this line has the common property that the weighted sum of oxidation states of the elements amounts to zero ( $\text{O}(\text{Zn})=+II$ ,  $\text{O}(\text{Al})=+III$ ,  $\text{O}(\text{O})=-II$ ). None of these 16 structures is stable against decomposition into ZnO,  $\text{Al}_2\text{O}_3$  and/or  $\text{ZnAl}_2\text{O}_4$ . Their excess formation enthalpy is sufficiently low, however, to expect that they could be stabilized e.g. as epitaxial thin films.

We plan to use an evolutionary search as a tool to identify crystal structures for Al-Zn-O stoichiometries along this line of interest, for which no analogies in chemically similar ternary systems are known.

### References

1. G. Bergerhoff, M. Hundt, R. Sievers, I.D. Brown, *J. Chem. Inf. Comput. Sci.*, **1983**, 23, pp 66–69.
2. G. Hautier, C.C. Fischer, A. Jain, T. Mueller, G. Ceder, *Chem. Mat.*, **2010**, 22(12), pp 3762–3767.
3. J. Carrasco, F. Illas, S.T. Bromley, *Phys. Rev. Lett.*, **2007**, 99, pp 235502.
4. M.A. Zwijnenburg, F. Illas, S.T. Bromley, *Phys. Rev. Lett.*, **2010**, 104, pp 175503.
5. G. Kresse, J. Hafner, *Phys. Rev. B*, **1993**, 47, pp 558.